

SYNTHESIS OF 2-(DIMETHYLLALLYL)-N-HYDROXYTRYPTOPHANS FROM INDOLE

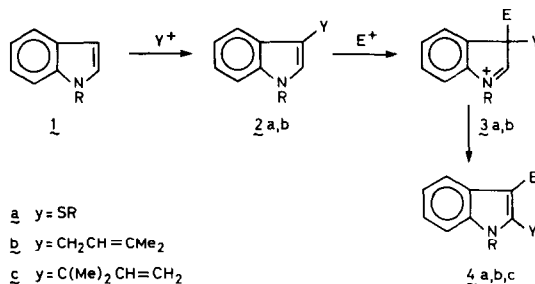
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Abstract: A mild and general method for the synthesis of dimethylallyl tryptophan derivatives **11** and **12** is described. Reaction of the 3-(α,α -dimethyl-allyl) indoles **7a** or **7b** with the nitroso olefin **8** yields the cycloadducts **9a** (93%) and **9b** (87%), respectively. Subsequent treatment of **9a** with trifluoroacetic acid affords the 2-(dimethylallyl) tryptophan derivatives **11a** and **12a** in a 9:1 ratio (97% yield). Isomerisation of **9b** gave only **11b** (88% yield).

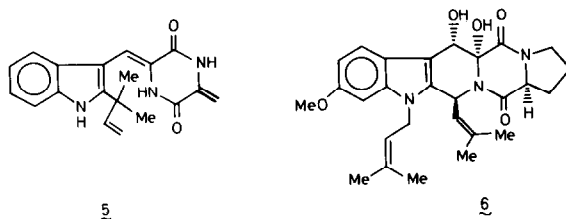
We have recently described a general method for the preparation of 2-(alkylthio) tryptophans and the corresponding N-hydroxy derivatives from indoles¹. The concept used is based on the reaction of indoles having an alkylthio substituent at C(3), e.g. **2a**, with an electrophile, viz. a nitroso olefin derivative (Scheme I). This reaction yields **4a** in which the alkylthio group has migrated from C(3) to C(2)². This approach has been exemplified³ by a synthesis of tryptathionine, the characteristic structural element of toxic principles of members of the mushroom *Amanita Phalloides*⁴. Subsequently we became intrigued by the potential usefulness of this scheme for the synthesis of 2-(dimethylallyl)-indole derivatives **4b-c**.

scheme I



Several mould metabolites have been isolated featuring a dimethylallyl moiety at C(2) of a tryptophan system. Examples having an α,α -dimethylallyl or an γ,δ -dimethylallyl moiety are Neoechinulin B (**5**)⁵ and Fumitremorgin B (**6**)⁶.

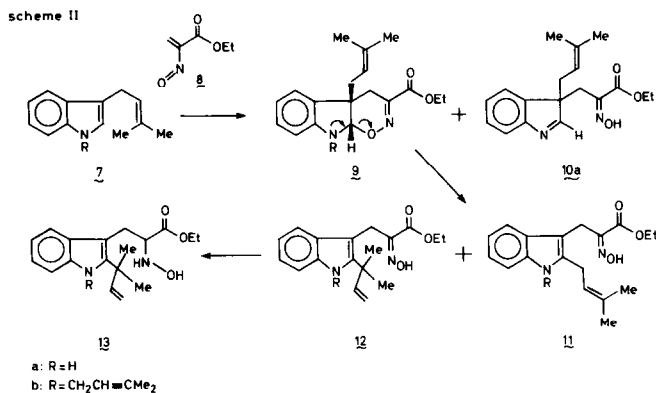
Figure



Various attempts to synthesize indole alkaloids featuring one of these dimethylallyl moieties at the C(2) position of tryptophan have been made⁷. Most satisfactory is the prenylation of tryptophan derivatives⁸, which has been used in a biomimetic approach to 6. However, no approach starting from indole (1) has been reported. Such an approach seemed attractive to us as it might allow facile introduction of substituents in the indole nucleus prior to the introduction of the amino acid side chain and the dimethylallyl moiety. Several of the above mentioned alkaloids *e.g.* 6 have such a substituent in the indole nucleus.

The purpose of this letter is to show that the 2-(dimethylallyl)-indoles 4b,c can be prepared indeed from 2b as depicted schematically in Scheme I. The indoles prepared have as an additional feature the presence of an N-hydroxy- α -amino acid side chain at C(3), which renders these compounds into versatile synthons⁹.

Treatment of 7a - prepared according to Casnati *et al.*¹⁰ - with the nitroso olefin 8 - prepared *in situ*⁹ from ethyl α -(hydroxyimino)- β -bromopropanoate (1.1 equiv., dichloromethane, 24 h, 20 °C, argon atmosphere) - gave a 2:1 mixture of the cycloadduct 9a¹¹ and its ring-opened isomer 10a¹² in 93% overall yield (Scheme II).



Separation of the isomers could be achieved by flash column chromatography (Silicagel 60, dichloromethane/n-hexane, 1/1, v/v). When 7b¹³ was subjected to these reaction conditions only cycloadduct 9b¹⁴ was formed (87% yield). Subsequently, isomerisation was achieved using trifluoroacetic acid (5 equiv., dichloromethane, 1 h, 20 °C). Compound

9a gave the compounds 11a¹⁵ and 12a¹⁶ in a 9:1 ratio (97% overall yield). These two isomers could be separated by crystallisation (dichloromethane/n-hexane). The same result was obtained when a mixture of 9a and 10a was used. Isomerisation of 9b afforded 11b¹⁷ (88% yield); in this case 12b could not be detected.

The formation of 11 and 12 can be rationalized as follows. The protonated form of 9 is in equilibrium with its ring-opened tautomer, an indoleninium derivative. The latter species might undergo either a C(3)-C(2) allyl shift followed by rearomatization to afford 11 or a Cope-like rearrangement to give 12 subsequent to rearomatization.

Finally, the oximino group of 12a was reduced selectively to an N-hydroxyamino function to afford 13a¹⁸ (79% yield, trimethylamine borane and HCl in ethanol⁹).

We are currently applying 11a and 13a in projected syntheses of indole alkaloids. One such study describes the synthesis of 5.

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11. Compound 9a (oil): EIMS: exact mass calcd. for C₁₈H₂₂N₂O₃; [M]⁺ 314.1630, found m/e 314.1634. ¹H-NMR (90 MHz, CDCl₃), δ=7.14-6.46 (m, 4H, indoline C(4)-C(7)H), 5.32 (s, 1H, indoline C(2)H), 5.14 (m, 1H, C=CH), 4.90 (s(br), 1H, indoline NH), 4.25 (q, 2H, OCH₂CH₃), 3.00 and 2.70 (AB spectrum, ²J_{AB}=15.8 Hz, 2H, indoline C(3)CH₂),

- 2.40 (d, $^3J=7.6$ Hz, 2H, indoline C(3)-CH₂-CH=), 1.72 (s, 3H, CH₃), 1.56 (s, 3H, CH₃), 1.29 (t, 3H, OCH₂CH₃).
12. Compound 10a (oil): EIMS: exact mass calcd. for C₁₈H₂₂N₂O₃ [M]⁺ 314.1630, found m/e 314.1620. ¹H-NMR (60 MHz, CDCl₃), δ=7.1-6.4 (m, 4H, indolenine C(4)-C(7)H), 5.5 (s, 1H, indolenine C(2)H), 5.1 (m, 1H, C=CH), 4.20 (dq, 2H, OCH₂CH₃), 3.1 and 2.5 (AB spectrum, $^2J_{AB}=15$ Hz, 2H, indolenine C(3)CH₂), 2.4 (d, $^3J=7.5$ Hz, 2H, indolenine C(3)-CH₂-CH=), 1.7 (s, 3H, CH₃), 1.5 (s, 3H, CH₃), 1.3 (t, 3H, OCH₂CH₃).
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14. Compound 9b (oil): EIMS: exact mass calcd. for C₂₃H₃₀N₂O₃ [M]⁺ 382.2256, found m/e 382.2264. ¹H-NMR (90 MHz, CDCl₃), δ=7.10-6.30 (m, 4H, indoline C(4)-C(7)H), 5.35-4.90 (m, 2H, 2x C=CH), 5.20 (s, 1H, indoline C(2)H), 4.20 (dq, 2H, OCH₂CH₃), 3.95 (d, $^2J=7$ Hz, 2H, N-CH₂), 3.05 and 2.40 (AB spectrum, $^2J_{AB}=15$ Hz, 2H, indolenine C(3)CH₂), 2.35 (d, $^3J=7$ Hz, 2H, indoline C(3)-CH₂-CH=), 1.73 (s, 6H, 2xCH₃), 1.65 (s, 3H, CH₃), 1.55 (s, 3H, CH₃), 1.25 (t, 3H, OCH₂CH₃).
15. Compound 11a: m.p. 127-129 °C (CH₂Cl₂/n-hexane): CIMS: exact mass calcd. for C₁₈H₂₃N₂O₃ [M+]⁺ 315.1709, found m/e 351.1712. ¹H-NMR (90 MHz, CDCl₃), δ=7.80 (s(br), 1H, indole NH), 7.74-7.00 (m, 4H, indole C(4)-C(7)H), 5.32 (m, 2H, C=CH), 4.22 (q, 2H, OCH₂CH₃), 4.04 (s, 2H, indole C(3)-CH₂), 3.60 (d, $^3J=7.3$ Hz, 2H, C=CH-CH₂), 1.75 (s, 6H, 2xCH₃), 1.26 (t, 3H, OCH₂CH₃). Anal. calcd. for C₁₈H₂₂N₂O₃ (M=314.385) C, 68.77; H, 7.05; N 8.91; found C, 68.71, H 7.06; N 8.88.
16. Compound 12a: m.p. 86-88 °C (CH₂Cl₂/n-hexane): CIMS: exact mass calcd. for C₁₈H₂₃N₂O₃ [M+]⁺ 315.1709, found m/e 315.1715. ¹H-NMR (90 MHz, CDCl₃), δ=9.20 (s(br), 1H, NOH), 8.90 (s, 1H, indole NH), 7.60-6.90 (m, 4H, indole C(4)-C(7)H), 6.19 (dd, $^3J_{trans}=17.6$ Hz, $^3J_{cis}=10.5$ Hz, 1H, CH=CH₂), 5.19 and 5.16 (2x dd, $^3J_{trans}=17.6$ Hz, $^3J_{cis}=10.5$ Hz, $^2J_{gem}=1.2$ Hz, 2H, CH=CH₂), 4.14 (q, 2H, OCH₂CH₃), 1.58 (s, 6H, C(CH₃)₂), 1.12 (t, 3H, OCH₂CH₃). Anal. calcd. for C₁₈H₂₂N₂O₃•1/6 CH₂Cl₂ (M=328.539) C, 66.42; H, 6.85; N 8.53; found C, 66.50, H 6.88; N 8.55.
- 12a was found to be identical to a product obtained by an alternative route, see Plate, R.; Ottenheijm, H.C.J. manuscript submitted.
17. Compound 11b: m.p. 138-142 °C (CH₂Cl₂/n-hexane): EIMS: exact mass calcd. for C₂₃H₃₀N₂O₃ [M]⁺ 382.2256, found m/e 382.2264. ¹H-NMR (90 MHz, CDCl₃), δ=8.50 (s(br), 1H, indole NOH), 7.72-6.87 (m, 4H, indole C(4)-C(7)H), 5.40 (m, 2H, 2x C=CH), 4.60 (d, $^3J=7$ Hz, 2H, N-CH₂-CH=), 4.17 (q, 2H, OCH₂CH₃), 4.03 (s, 2H, indole C(3)-CH₂), 3.56 (d, $^3J=7$ Hz, 2H, indole C(2)-CH₂-CH=), 1.92 and 1.77 (2xs, 12H, 4xCH₃), 1.25 (t, 3H, OCH₂CH₃). Anal. calcd. for C₂₃H₃₀N₂O₃ (M=382.504) C, 72.22; H, 7.91; N 7.32; found C, 71.90, H 7.90; N 7.22.
18. Compound 13a: (foam). EIMS: exact mass calcd. for C₁₈H₂₄N₂O₃ [M]⁺ 316.1787, found m/e 316.1784. ¹H-NMR (90 MHz, CDCl₃), δ=8.90 (s(br), 1H, indole N'H'), 7.60-7.00 (m, 4H, indole C(4)-C(7)H), 6.14 (dd, $^3J_{trans}=17.6$ Hz, $^3J_{cis}=10.2$ Hz, 1H, CH=CH₂), 5.50 (s(br), 2H, NHOH), 5.28 and 5.13 (2x dd, $^3J_{trans}=17.6$ Hz, $^3J_{cis}=10.2$ Hz, $^2J_{gem}=1.2$ Hz, 2H, CH=CH₂), 4.16 (q, 2H, OCH₂CH₃), 4.02 (X part of ABX spectrum, $^3J_{AX}=6.3$ Hz, $^3J_{BX}=9.0$ Hz, 1H, CH₂-C'H'), 3.21 and 3.12 (AB part of ABX spectrum, $^3J_{AX}=6.3$ Hz, $^3J_{BX}=9.0$ Hz, $^2J_{AB}=15.3$ Hz, 2H, CH₂-CH), 1.56 (s, 6H, C(CH₃)₂), 1.18 (t, 3H, OCH₂CH₃).

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